Ab initio studies of the electronic structure of defects in PbTe

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Understanding the detailed electronic structure of deep defect states in narrow band-gap semiconductors has been a challenging problem. Recently, self-consistent *ab initio* calculations within density functional theory using supercell models have been successful in tackling this problem. In this paper, we carry out such calculations in PbTe, a well-known narrow band-gap semiconductor, for a large class of defects: cationic and anionic substitutional impurities of different valence, and cationic and anionic vacancies. For the cationic defects, we study the chemical trends in the position of defect levels by looking at series of compounds $RPb_{2n-1}Te_{2n}$, where *R* is vacancy or monovalent, divalent, or trivalent atom. Similarly, for anionic defects, we study compounds $MPb_{2n}Te_{2n-1}$, where *M* is vacancy, S, Se or I. We find that the density of states near the top of the valence band and the bottom of the conduction band get significantly modified for most of these defects. This suggests that the transport properties of PbTe in the presence of impurities may not always be interpreted by simple carrier doping (from bound impurity states in the gap) concepts, confirming such ideas developed from qualitative and semiquantitative arguments.

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I. INTRODUCTION

Narrow band-gap IV-VI binary semiconductors and their alloys have been of great interest because of their fundamental electronic properties and for their practical applications. Lead chalcogenide salts PbTe and PbSe are two IV-VI narrow-gap semiconductors whose studies over the last several decades have been motivated by their importance in infrared detectors, light-emitting devices, infrared lasers, thermophotovoltaics, and thermoelectrics.^{1–3} In fact, PbTe was one of the first materials studied by Ioffe and his colleagues in the middle of the past century when there was a revival of interest in thermoelectricity.⁴ This compound, its alloys with SnTe and PbSe, and related compounds called TAGS (alloys of AgSbTe₂ and GeTe) were for many years the best thermoelectric materials at temperatures \sim 700 K.⁵ In recent years, quantum wells of PbTe/Pb_{1-x}Eu_xTe, PbSe_{0.98}Te_{0.02}/PbTe superlattices⁶ and quaternary compounds $AgSbPb_{2n-2}Te_{2n}$ $(n=9,10)^7$ have attracted considerable attention because of their large thermoelectric figure of merit (FOM). Similar quaternary systems with Ag replaced by Na are also promising high temperature thermoelectrics.⁸ Most of the above systems have stoichiometry closer to the parent compound PbTe.

It is well known that transport and optical properties of semiconductors are dominated by the states in the neighborhood of the band gap. For example, in thermoelectrics, the FOM (denoted as *ZT* where *T* is the operating temperature) depends on the thermopower *S* and electric conductivity σ through the relation $ZT = \sigma S^2 T / \kappa$, where κ is the sum of electronic and lattice thermal conductivities of the material.⁹ Clearly, large values of *ZT* require large values of *S* and σ both of which depend sensitively on the nature of the electronic states near the band gap.¹⁰ Thus it is important to have a fundamental understanding of the changes in the electronic states near the band-gap region in PbTe when it is mixed with other binary and ternary compounds. Before being able

to understand the nature of the electronic states near the band gap and/or the Fermi energy in the complex systems mentioned above, one must understand how the electronic states of PbTe change when Pb or Te is replaced by impurity atom or by vacancy.

A detailed microscopic understanding of the nature of the defect states in PbTe is an old but a challenging problem. Naively, one expects to see bound shallow defect states in the band gap [pinned to the valence band maximum (VBM) or conduction band minimum (CBM)] when a Pb or a Te atom is replaced by either a donor- or acceptor-like impurity or by a vacancy. However, it is known that, in narrow bandgap semiconductors such as PbTe, one does not see these shallow defect states but deep defect states which can be outside the band gap.^{11–15} Theoretically, a fundamental understanding of the shallow defect states in semiconductors with large band gap traces back to the classic work of Kohn and Luttinger.¹⁶ In this case, bound defect states are produced by the long-range Coulomb potential of charged defects screened by the dielectric constant of the host semiconductor. These states appear below the CBM (for donor impurities) or above the VBM (for acceptor impurities). Their binding energies are well described by a hydrogenic model after taking proper account of the band effective mass, dielectric screening, and a small central-cell correction.¹² However, even in large band-gap semiconductors, one can have deep defect states (strongly bound) when the centralcell effects are equally important as the one caused by the Coulomb potential.¹² Unlike the shallow impurity levels, the deep levels (in narrow band-gap semiconductors) are dominated by the short-range atomiclike defect potential and changes in local bonding. Consequently, the detailed physics underlying the formation of deep defects states in these systems depends sensitively on the specific features of the defect.

One of the earlier attempts to understand the origin of deep defect states in PbTe was made by Lent *et al.*¹³ For substitutional defects in PbTe, they presented a simple

chemical theory for a large class of substitutional defects and vacancies in terms of the atomic energy levels and tight binding concepts. Unfortunately, these calculations did not take into account the potentials generated by the impurities in a self-consistent manner and, therefore, were only qualitative in their prediction. In fact, an even earlier attempt to understand the origin of vacancy-induced deep defect states in PbTe was made by Parada and Pratt.¹⁴ They used a Slater-Koster model and Wannier function basis constructed out of a finite number of PbTe bands. They predicted defect states outside the band gap and were able to explain the low temperature *n*- and *p*-type transport properties in nominally pure PbTe. The defect potential used in the Slater-Koster model was assumed to be a constant and was not calculated selfconsistently. Therefore the precise positions of the defect states predicted by this calculation are not expected to be quantitatively reliable. The other major efforts in understanding the physics of deep defect states are for In-doped PbTe. Inhomogeneous mixed-valence models for In and other group III impurities have been proposed to explain several experimental data.¹⁷ In these models, In impurities exist in two valence states, one trivalent and the other monovalent, stabilized by strong electronic and ionic relaxations. Our recent calculations have cast some doubts on the validity of this model.¹⁸

In this paper, we discuss the results of our attempts to understand the physics behind the deep defect states in a large class of defects in PbTe using self-consistent ab initio electronic structure calculations within density functional theory (DFT). Only a few ab initio studies have been made recently. One was by Bilc et al.,¹⁹ who recently reported their results on Ag and Sb substitutional defects in PbTe using all-electron full-potential linearized augmented plane-wave (FP-LAPW) method.²⁰ Another was by Ahmad *et al.*¹⁸ who looked at the In impurity problem. Similar calculations have been carried out by Edwards et al.,²¹ who looked at the Ge and Te vacancy and antisite defects in GeTe, and by Lany et al^{22} on the DX centers in CdTe. The purpose of our present work is to extend the previous studies of the defects in PbTe to many other different defects: Pb and Te vacancies, Pb substituted by monovalent atoms (alkalis, Cu), divalent atoms (s-type Zn, Cd, Hg, and p-type Ge, Sn), or trivalent atoms (group III Ga, In, Tl, and group V As, Sb, Bi), and Te substituted by S, Se, or I. Our main focus is to develop a systematic understanding of these defect states as one goes across or down the periodic table to select different impurity atoms. Preliminary results of this study have been recently reported.^{23,24} The arrangement of this paper is as follows. In Sec. II, we describe the structural model and discuss briefly the method used to calculate the electronic structure. In Sec. III, we present our calculation results focusing on the electronic density of states (DOS) with and without the defects. We give the change in the total and partial DOS caused by different types of defects and discuss the underlying physics. In Sec. IV, we discuss our theoretical results in the light of recent measurements of thermopower in several PbTe based compounds. Finally, we give a brief summary in Sec. V.

II. METHOD OF CALCULATION

Before we discuss the structural model used to study the impurity states, we would like to briefly describe the quaternary systems one constructs in developing thermoelectrics starting from the binary compound PbTe and replacing Pb atoms by other atoms. The compounds are described by the formula $RR'Pb_{2n-2}Te_{2n}$, where \overline{R} and R' are chosen to maintain charge neutrality. Here two divalent Pb atoms are replaced by one monovalent (R) and one trivalent (R') atom. If R is divalent then R' is also divalent (for example R=R'=Cd). The material is therefore charge compensated and most likely will maintain its semiconducting character. One then adjusts the concentration of R and R' to control the sign of the charge carriers and optimize their concentration. In a recent work, Bilc et al.¹⁹ have discussed the electronic structures of a series of charge compensated AgSbPb_{2n-2}Te_{2n} compounds (n=6,9,16) and also charge uncompensated compounds $AgPb_{2n-1}Te_{2n}$, $SbPb_{2n-1}Te_{2n}$ (n=16) where only one of the elements R or R' is present. They found resonant states overlapping conduction and valence bands caused by Sb and Ag impurities, respectively. In this paper, we focus on defects of one type (either R or R') and investigate the electronic structure of $RPb_{2n-1}Te_{2n}$. We also carry out similar calculations in compounds of the type $MPb_{2n}Te_{2n-1}$, where *M* is a defect at the Te site. Our aim is to explore systematically the impurity-induced changes in the electronic DOS as one goes down or across the periodic table.

We use a supercell model where the defects are periodically arranged in a PbTe lattice. This corresponds to the formula $RPb_{2n-1}Te_{2n}$ or $MPb_{2n}Te_{2n-1}$ as mentioned above. To increase the distance between the defects as much as possible (in order to reduce the interaction between them) without increasing the size of the supercell enormously we have chosen n=16. This corresponds to a unit cell containing 64 atoms with the minimum distance between two defects being \sim 13 Å (see Fig. 1). Due to the large dielectric constant of the host PbTe, we expect long-range Coulomb effects to be strongly screened, thereby reducing the impurity-impurity interaction. However, short-range interactions are still present. Before describing the details of our calculations we would like to point out that there are some fundamental problems regarding the inadequacy of using the supercell model to calculate the energy of isolated defect states in semiconductors;²⁵ however, they are usually for extended



FIG. 1. (Color online) Supercell model of $RPb_{31}Te_{32}$ where *R* is either a vacancy or an impurity atom. Small balls are for Pb and Te, large balls are for *R*.

"shallow" impurity states. Since most of the defect states in narrow band-gap semiconductors of our interest are of strongly localized "deep" type, we expect the supercell model to be reasonable. We discuss in details the supercell size dependence of the defect states in Sec. III C of the paper.

Electronic structure calculations were performed within DFT using all-electron FP-LAPW plus local orbital method²⁰ incorporated in WIEN2K.²⁶ We used the generalized-gradient approximation (GGA) (Ref. 27) for the exchange and correlation potential. Scalar relativistic corrections were included and spin-orbit interaction (SOI) was implemented using a second variational procedure.²⁸ Convergence of the selfconsistent iterations was performed using 20 k points inside the reduced Brillouin zone to within 0.0001 Ry (1.36 meV) with a cutoff energy of -6.0 Ry separating the valence and the core states. For this initial survey, calculations were performed using the experimental lattice constant of PbTe, which is 6.454 Å,²⁹ and no relaxation (volume or internal) studies were carried out. For a few systems, we also carried out calculations using projector augmented wave (PAW) (Ref. 30) method incorporated in Vienna *ab initio* Simulation Package (VASP) (Ref. 31) to compare with the FP-LAPW results and also to increase the impurities separation by going to larger supercells. The optimal lattice constant of PbTe obtained from volume optimization (which is 6.55 Å) was used in these PAW calculations, and ionic relaxations were performed. In our future work, we propose to carry out complete relaxation studies in all the defect systems.

III. RESULTS AND DISCUSSION

Before presenting the results of the *ab initio* electronic structure calculations, we discuss some basic physics of PbTe system. In this compound Pb and Te 6s and 5s states are quite deep and the valence and conduction bands are formed primarily out of 6p states of Pb and 5p states of Te. The Pb-Te bonds have both covalent and ionic character. The removal of one Pb atom or replacing it by a monovalent cation leads to a local charge disturbance and alters the covalent bonding with the neighboring Te (denoted as Te2) pstates. As a result, the valence band states get strongly perturbed. One expects acceptorlike states to appear near the top of the valence band. However, their precise positions can only be obtained after a detailed self-consistent calculation. Also, because of the local nature of the perturbation we expect the entire valence band region to be affected. In case of the monovalent Ag impurity, one has, in addition, the filled dstates which interact with the Te p states and change the valence band. The Cd impurity, although divalent like Pb, has 5s state which may give rise to states near the bottom of the conduction band. Indium (In) and other trivalent sp atoms may also give rise to a bound s state below the valence band and therefore act more like a monovalent impurity than a trivalent one. It is important to emphasize that simple shallow impurity-state models¹⁶ or Slater-Koster type models calculating without the impurity potential self-consistently^{13-15,17} are not capable of describing the short-range physics of these impurity states quantitatively. Only a careful self-consistent ab initio calculation will be able to do so. In this paper we have focused primarily on the cationic site defects. This is because in some experimental systems of our interest, e.g., $AgSbPb_{2n-2}Te_{2n}^{7}$ and $Na_{0.95}Pb_mSbTe_{m+2}$,⁸ the Ag, Na, and Sb atoms are located on the Pb-cation sites. Furthermore, this system is an *n*-type semiconductor and since electrons in PbTe move in predominantly Pb-based orbitals, we are interested in the effects of cationic dopants on the DOS. Although anionic vacancies have been considered to play a significant role in the transport properties of PbTe, substitutional defects at the anionic sites have been less well investigated in the past and there are plans for future investigation of these defects.

A. Pb-site defects

The removal of one Pb atom (i.e., creation of one vacancy at the Pb site) or replacing it by another atom with different valence introduces a local charge disturbance and alters the bonding with the neighboring Te p states. Due to the longrange nature of the former and the short-range of the latter, we expect the entire valence and conduction band states to get perturbed in the presence of the impurity.

1. Pb vacancy in PbTe

The total DOS of PbTe with and without Pb vacancy and the difference between the two are shown in Figs. 2(a)-2(c). For the sake of comparison, we shifted the DOS of PbTe by a small amount ($\sim 0.2 \text{ eV}$) so that the lower edge of the valence band DOS of the systems with and without Pb vacancy matched. We found that the first peak below the valence band maximum (VBM) also matched after this shift. This small shift appears to be reasonable since with this shift the Te core 5s bands also matched perfectly. As we can see from Figs. 2(a)-2(c), the entire valence band and conduction band DOS get modified by the presence of the Pb vacancy. It is important to note that the states near the bottom of both the valence and conduction band do not change appreciably in the presence of the Pb vacancy. Since transport properties are controlled by the states near the VBM and conduction band minimum (CBM), let us focus on the band-gap region within an energy range of ± 0.5 eV about the center of the band gap. We see that there is very little change near the bottom of the conduction band DOS whereas there is an increase in the DOS near the top of the valence band. We analyze this increase in more detail below. From Fig. 2(a), we can clearly see that the Pb vacancies give rise to holes as expected.

To see which atoms contribute to the change in the DOS near the VBM, we show in Fig. 2(b) the difference in the partial DOS between PbTe with and without Pb vacancy. There are six nearest-neighbor Te atoms of the Pb vacancy (denoted as Te2) and three other inequivalent Te atoms (denoted as Te1, Te3, and Te4). Among the Pb neighbors of the Pb vacancy, Pb2 is the Pb atom that bridges two Te2 atoms each being the nearest neighbor of one of the defect sites. In one supercell there are three Pb2 atoms and 28 other Pb atoms. Of the latter, there are four inequivalent (Pb3, Pb4, Pb5, Pb6) lead atoms. Clearly the major contribution to the DOS increase near the top of the valence band comes from



FIG. 2. (a) The total DOS of PbTe with and without Pb vacancy, (b) the difference in the partial DOS between PbTe with and without Pb vacancy where the six nearest-neighbor Te atoms of the Pb vacancy are denoted as Te2 and there are three other inequivalent Te atoms (denoted as Te1, Te3, and Te4), and (c) the difference in the total DOS ($\Delta \rho_{tot}$), in the DOS associated with the charges inside the atomic spheres ($\Delta \rho_{at}$), and in the DOS associated with interstitial regions ($\Delta \rho_i = \Delta \rho_{tot} - \Delta \rho_{at}$). The energy origin in (b) and (c) is chosen to be the top of the valence band of PbTe; whereas the energy origin in (a) is the highest occupied Kohn-Sham single-particle state (which will be denoted as the Fermi energy) of the system with defect.

the six nearest-neighbor Te2 atoms and we see that Te2 states in the lower half of the valence band get pushed towards the upper half. This transfer of the Te2 DOS is partially compensated by the downward shift of the DOS associated with the other atoms. Also other Te atoms do contribute a small amount to the increased DOS near the VBM.

It should be noted that, in FP-LAPW calculations, there is an appreciable DOS associated with charges outside the atomic spheres, i.e., in the interstitial space. In Fig. 2(c), we show the difference in the total DOS ($\Delta \rho_{tot}$), in the DOS associated with the charges inside the atomic spheres ($\Delta \rho_{at}$), and in the DOS associated with the interstitial space ($\Delta \rho_{i}$), $=\Delta \rho_{tot} - \Delta \rho_{at}$). $\Delta \rho_i$ and $\Delta \rho_{at}$ have almost the same energy dependence and are comparable in magnitude. We can interpret the peak in $\Delta \rho_{tot}$ near the top of the valence band [Fig. 2(c)] as a vacancy-induced resonant state. This should be significant in the hole-doped samples where these states will be involved in charge and energy transport. The fact that



FIG. 3. The difference in the total DOS ($\Delta \rho_{tot}$) between PbTe with and without monovalent impurities: Na, K, Rb, Cs, Cu, and Ag. The energy origin (0 eV) is chosen to be at the top of the valence band of PbTe.

Pb2, Te1, Te3, and Te4 contribute very little to $\Delta \rho_{tot}$ suggests that the vacancy-induced resonant states are weakly interacting.

It is interesting to compare our results with those of Parada and Pratt.¹⁴ They found that the perturbation due to a Pb vacancy was not strong enough to drive any levels out of the Te p bands. The Pb s state associated with the vacancy site was driven about 3 eV above the bottom of the conduction band, hence of no significance for transport. A doubly degenerate state was driven out of the Te s band but by about 1 eV, hence it was occupied by two Te s electrons. As a result, a Pb vacancy creates two holes in the Te p valence band and leads to p-type conduction. These findings are supported by our *ab initio* calculations. However, the subtle ef-



FIG. 4. The difference in the total DOS ($\Delta \rho_{tot}$) between PbTe with and without *s*-type divalent atom: Zn, Cd and Hg. The energy origin (0 eV) is chosen to be at the top of the valence band of PbTe.



FIG. 5. (a) The total DOS of PbTe with and without Cd impurity, (b) their difference $\Delta \rho_{\text{tot}}$, (c) partial DOS per Te associated with nearest-neighbor Te atoms, and (d) Cd partial DOS of *s*-character. The Fermi energy of PbTe with Cd impurity is at the energy origin (0 eV).

fects such as the DOS enhancement near the top of the Te p valence band which can affect the thermopower of holes was not investigated by them.

2. Monovalent impurities (Na, K, Rb, Cs, Cu, Ag) in PbTe

Now let us discuss the monovalent impurities. They should provide a weaker perturbation compared to the vacancy. In Fig. 3, instead of giving the total DOS, we only show the difference ($\Delta \rho_{tot}$). We find that K, Rb, Cs, Cu, and Ag all show a peak near the top of the valence band, just like a vacancy. Na is anomalous; it does not seem to change the DOS near the valence band top, but shows an increase (as compared to other three alkalis) near -3.2 eV. The major difference between Cu, Ag, and the alkalis is seen near the bottom of the valence band in the energy range (-2,-1) eV for Cu and (-4, -3) eV for Ag. This is due to the Cu and Ag d states hybridizing with the Te p bands. We will further discuss these results later in Sec. IV when we compare our theoretical results with experiment. We find that the valence *s* state of the alkalis are all above the valence band; hence, all of them donate one electron to the system. In contrast to the suggestions made in Ref. 13, we find that Cs behaves just like Na in this respect. We should note that even if the alkali atoms donate one electron when they replace Pb (which is divalent), they give rise to one hole per impurity and hence act as acceptors.

3. Divalent impurity (Zn, Cd, Hg, Sn, Ge) in PbTe

There are two types of divalent impurities: *s* type (Zn, Cd, and Hg) and *p* type (Sn and Ge). The valence configurations of the *s*-type impurities are $4s^2$ (Zn), $5s^2$ (Cd), and $6s^2$ (Hg); whereas those for the *p* types are $4p^2$ (Sn) and $5p^2$ (Ge). The

valence state of divalent Pb is, of course, $6p^2$. We find that the *s*-type impurities introduce resonant states near the bottom of the PbTe conduction band and there are three major changes in the DOS: (i) an enhancement of the DOS near the bottom of the conduction band caused by the resonant state, (ii) a reduction in the DOS near the top of the valence band within an energy range ~0.25 eV, and (iii) an enhancement of the DOS near the bottom of the valence band [see Fig. 4 and Figs. 5(a)-5(d)]. These three changes can be seen clearly if, for example, one looks at the Cd *s* character in Fig. 5(d). The Cd atom introduces an *s*-like state near the bottom of the conduction band. This state strongly hybridizes with the *p*



FIG. 6. The difference in the total DOS ($\Delta \rho_{tot}$) between PbTe with and without *p*-type divalent atom: Sn and Ge. The energy origin (0 eV) is chosen to be at the top of the valence band of PbTe.



FIG. 7. The difference in the total DOS ($\Delta \rho_{tot}$) between PbTe with and without trivalent group III impurities: Ga, In, and Tl. The energy origin (0 eV) is chosen to be at the top of the valence band of PbTe.

orbitals of the six neighboring Te2 atoms and the bonding state appears at about -4 eV [see Figs. 5(c) and 5(d)]. The corresponding antibonding state appears as a resonant state near the conduction band bottom, reflecting the deep defect nature of these impurity states. The states near the top of the valence band (antibonding between Pb and Te2 *p*'s) are depressed when Pb is replaced by Cd. Because of the resonant state near the bottom of the conduction band, we expect electron-doped Cd systems to exhibit large thermopower. Its experimental implication will be discussed later. In contrast to the *s*-type divalent impurities, the *p*-type impurities, Sn and Ge, do not alter the DOS near the band gap (see Fig. 6). They behave very much like Pb near the band-gap region, which is understandable since Pb is also a p-type divalent atom.

4. Trivalent impurity (Ga, In, Tl, As, Sb, Bi) in PbTe

Like the divalent case, we also have two types of trivalent impurities: group III (Ga, In, Tl) and group V (As, Sb, Bi). The valence configurations of the group III trivalent atoms are $4s^2 4p^1$, $5s^2 5p^1$, and $6s^2 6p^1$ for Ga, In and Tl, respectively; whereas the valence configurations of the group V trivalent atoms are $4p^3$, $5p^3$, and $6p^3$ for As, Sb and Bi, respectively. All three group III impurities show a strongly bound state below the bottom of the valence band and a state near the band-gap region. These strongly bound states can be identified with the electrically inactive "hyperdeep" levels proposed by Hjalmarson *et al.*³² which always appear along with the electronically active deep defect states in the bandgap region. These deep defect states are seen in our calculation as peaks in $\Delta \rho_{tot}$ as shown in Fig. 7 near the Fermi energy (at 0 eV).

Results for Ga and Tl atoms show that their deep defect states are resonant states near the top of the PbTe valence band (Fig. 7), whereas for In this state lies in the band-gap region.¹⁸ This difference between In and the other two may explain the experimental anomalies seen in the case of In impurities in PbTe.¹⁷ The resonant states at the top of the valence band for the Ga and Tl is caused by the hybridization of the *s* states of these atoms with the *p* state of Te2. We show, in Figs. 8(a)–8(d), the total and partial DOS of PbTe with and without Ga impurity, the difference $\Delta \rho_{tot}$, and *s* character of the partial DOS of Ga impurity. Results for Tl impurity are similar excepting that the hyperdeep level is



FIG. 8. (a) The total DOS of PbTe with and without Ga impurity, (b) their difference $\Delta \rho_{\text{tot}}$, (c) partial DOS per Te associated with nearest-neighbor Te atoms, and (d) Ga partial DOS of *s* character. The Fermi energy of PbTe with Ga impurity is at the energy origin (0 eV).



FIG. 9. The difference in the total DOS ($\Delta \rho_{tot}$) between PbTe with and without trivalent group V impurities: As, Sb, and Bi. The energy origin (0 eV) is chosen to be at the top of the valence band of PbTe.

further below the valence band minimum. The hyperdeep level at about -5.0 eV is predominantly impurity(Ga)-like whereas the resonant state centered around 0 eV is predominantly host (Te2)-like, which is consistent with the picture proposed by Hjalmarson *et al.*³² The peak of the resonant state is about 0.25 eV below the VBM. This is consistent with the current understanding of the Tl-doped PbTe systems (see Ref. 17 and references therein). According to our calculations, Ga and Tl impurities should show rather similar behavior.

Indium (In) is different from Ga and Tl. The deep defect state lies in the band gap and hence localized. This localized state is occupied by one electron. According to the mixedvalence model of In impurities in PbTe,¹⁷ two In impurities substituting for two Pb atoms give two electrons each to the Te *p* band leaving them as two In²⁺ ions. However, a pair of In²⁺ ions undergo charge disproportionation and become In³⁺ and In¹⁺, resulting in an inhomogeneous mixed-valence state for the pair. In our calculations, we see that to create an In³⁺ ion will require taking two electrons from the hyperdeep levels to the top of the valence band and will cost too much energy. We have checked this by taking two In impurities in our 64-atom unit cell and allowing for lattice relaxation. We do not find any evidence for the presence of an In³⁺ state.¹⁸

Results for group V impurities (As, Sb, and Bi) show that they introduce resonant states near the bottom of the conduction band (see Fig. 9). In contrast to the group III impurities which have s and p valence electrons, the resonant states for these p valence electron impurities are rather broad, the sharpest structure being for As. In Figs. 10(a)-10(d), we show the total and partial DOS of PbTe with the As impurity. The resonant state is a mixture of the p states of As and the p states of the neighboring Te2 atoms in almost equal proportions.

B. Te-site defects

As in the case of Pb-site defects, the removal of one Te atom (i.e., creation of one vacancy) or replacing it by other atom introduces a local charge disturbance and alters the covalent bonding with the neighboring Pb (denoted as Pb2) p states. As a result, we expect both the valence and conduction bands states to get strongly perturbed.



FIG. 10. (a) The total DOS of PbTe with and without As impurity, (b) their difference $\Delta \rho_{tot}$, (c) partial DOS per Te associated with nearest-neighbor Te atoms, and (d) As partial DOS of *p* character. The Fermi energy of PbTe with As impurity is at the energy origin (0 eV).



FIG. 11. (a) The total DOS of PbTe with and without Te vacancy, (b) their difference $\Delta \rho_{tot}$, and (c) partial DOS per Pb associated with nearest-neighbor Pb atoms. The Fermi energy of PbTe with Te vacancy is at the energy origin (0 eV).

1. Te vacancy in PbTe

The total and partial DOS of PbTe with and without vacancy and the difference $\Delta \rho_{tot}$ are shown in Figs. 11(a)–11(c). For the sake of comparison, we shifted the DOS of PbTe with defect by ~0.455 eV such that the core bands match perfectly. As we can see, the entire valence and conduction band DOS get perturbed by the vacancy. Since transport properties are controlled by the states near the Fermi energy (at 0 eV), let us focus on this region. As seen in Figs. 11(a)–11(c), additional states appear in the band-gap region and near the bottom of the conduction band in the energy range –0.5 to 0 eV. Partial DOS analysis indicates that these states are predominantly of Pb *p* character [see Fig. 11(c)]. The valence band loses six states (including spin) per Te vacancy. These six states are occupied by four Te electrons and two Pb electrons. Removal of Te atom takes four valence electrons away leaving two Pb valence electrons. As a result, two Pb electrons occupy the additional states mentioned above. The system then behaves like an *n*-doped system. Our picture is qualitatively consistent with that of Parada¹⁵ who argued that each Te vacancy causes four states (eight states including spin) to be removed from the valence band and to move near the conduction band bottom. This picture includes the Te 5s core state. Since Te atom has six electrons, the two electrons (coming from Pb) occupying the eight states in valence band (in PbTe) have to move into the conduction band resulting in an *n*-type conduction. However, the detailed nature of the DOS near the bottom of the conduction band could not be obtained in earlier non-self-consistent calculations.¹⁵

2. S and Se impurities in PbTe

S and Se, like Te, need two electrons to fill their valence p shell. Thus one expects mainly the valence band to be perturbed when one replaces Te by S or Se, and there should be a small perturbation of the conduction band (indirectly through changes in the hybridization effects). This is clearly seen in Figs. 12(a)–12(c) and Figs. 13(a)–13(c). The change in the valence band DOS takes place over the entire valence band energy range with a reduction in the DOS in the energy range of 0.5 eV from the top of the valence band. S and Se p states appear in the entire valence band region peaking around 1.2–1.5 eV below the VBM [Fig. 12(d) and Fig. 13(d)]. The conduction band is hardly affected within about 1.0 eV from its minimum.

3. I impurity in PbTe

Iodine (I) appears to act as an ideal donor in PbTe, in agreement with what is known about the halogen impurities in PbTe.^{17,33} It does not perturb the states both near the top of the valence band and near the bottom of the conduction band as can be seen in Figs. 14(a)-14(d). The major change in the DOS occurs near the bottom of the valence band around -5.0 eV (predominantly I *p* character), near -1.1 eV and +1.0 eV due to changes in the Pb *p* hybridization effect. The Fermi energy is shifted to the conduction band making it an *n*-type impurity (donor).

C. Impurity-impurity interaction and the dilute defect limit

In the calculation of energy of the defect states using the supercell model containing 64 atoms, the defect-defect interaction is strong. It is therefore important to understand how much of the defect physics remains as one increases the distance between the impurities (when one goes to the dilute limit). Clearly the spectral weight of the impurity-induced changes in the DOS will scale with impurity concentration. In addition, we expect the impurity bandwidth to decrease. But the most significant question is how the position of say the impurity band center changes relative to the conduction or valence band extrema. In order to see how the impurity-impurity interaction affects the DOS when two impurities are closer or farther than 13.16 Å, we have studied two different



FIG. 12. (a) The total DOS of PbTe with and without S impurity, (b) their difference $\Delta \rho_{tot}$, (c) partial DOS per Pb associated with nearest-neighbor Pb atoms, and (d) *S* partial DOS of *p* character. The Fermi energy of PbTe with S impurity is at the energy origin (0 eV).

systems. One is $In_2Pb_{31}Te_{32}$, i.e., two impurities in a $(2 \times 2 \times 2)$ supercell, where one In is located at the corner and the other at the center of the supercell, the distance between two In impurities being 11.345 Å. The second is $InPb_{107}Te_{108}$ with In located at the corner of a $(3 \times 3 \times 3)$ supercell, the distance between the two In impurities being 19.65 Å. For these two systems, we used the PAW (Ref. 30) method since it is more computationally efficient than FP-LAPW (Ref. 20)

in handling large supercells like the $(3 \times 3 \times 3)$ one containing 216 atoms. For the sake of clarity, we present here the results of calculations without the spin-orbit interaction (SOI), since the SOI simply makes the band gap smaller but does not change the physics we are going to discuss below.

In the case of $In_2Pb_{31}Te_{32}$, the bandwidth of the hyperdeep defect state is the same as that of $InPb_{31}Te_{32}$ (see Fig. 15). This can be understood using symmetry argument. Since



FIG. 13. (a) The total DOS of PbTe with and without Se impurity, (b) their difference $\Delta \rho_{tot}$, (c) partial DOS per Pb associated with nearest-neighbor Pb atoms, and (d) Se partial DOS of *p* character. The Fermi energy of PbTe with Se impurity is at the energy origin (0 eV).



FIG. 14. (a) The total DOS of PbTe with and without I impurity, (b) their difference $\Delta \rho_{\text{tot}}$, (c) partial DOS per Pb associated with nearest-neighbor Pb atoms, (d) I partial DOS of *p* character. The Fermi energy of PbTe with I impurity is at the energy origin (0 eV).

the impurity states have s symmetry and the dominant contribution to the overlap between two impurity states comes through Te-p and Pb-p orbitals, one expects the hopping amplitude between the two impurity states (in an effective tight-binding model) to be small due to 90° relative arrangement of the overlap paths connecting the two impurities. In the case of InPb₁₀₇Te₁₀₈, our main results are summarized in Table I and Fig. 16. As we had expected, both the hyperdeep and deep defect states get narrower since the overlap between the In atoms becomes smaller. The bandwidths (as measured by $E_2 - E_1$) change from 0.323 to 0.049 eV for the hyperdeep (a factor of 6.5 reduction) and from 0.654 to 0.120 eV for the deep defect state (a factor of 5.45 reduction). In order to see how the changes in the impurity separation affects their positions of the impurity band centers, we have used the bottom of the valence band (conduction band) as the reference point E_{0v} (E_{0c}) for the hyperdeep (deep), since the valence and conduction bands are expected to be less sensitive to the size of the supercell than the impurity states. As seen in Table I, E_{0v} changes by only 0.001 eV and E_{0c} by 0.036 eV in going from $(2 \times 2 \times 2)$ to $(3 \times 3 \times 3)$ supercell; the energy separation between the hyperdeep level and E_{0v} (as measured by $E_{av} - E_{0v}$) increases from 0.425 to 0.449 eV, a rather small change of 0.024 eV (5.6% change); and the corresponding separation between the deep state and E_{0c} (as measured by $E_{av}-E_{0c}$) increases from 0.511 to 0.525 eV, a yet smaller change of 0.014 eV (2.7%). We then speculate that if we increase the distance between the impurities further, the main effect will be to reduce the impurity bandwidths without further change in their positions with respect to the band minima. Therefore we can use the results of smaller supercell calculations to discuss the qualitative physics.

IV. COMPARISON WITH EXPERIMENT

One of the remarkable properties of lead chalcogenide group of narrow band-gap semiconductors is that they have a range of nonstoichiometry accompanied by either Pb or Te vacancies. Our theoretical results predict that Pb vacancies



FIG. 15. The total DOS of PbTe with In impurity obtained in PAW calculations using two different models: $InPb_{31}Te_{32}$ with one In impurity at the corner, and $In_2Pb_{30}Te_{32}$ with one In at the corner and the other one at the center of a $(2 \times 2 \times 2)$ supercell; the distance between the two In impurities being 13.1 Å and 11.345 Å, respectively. The two lines are almost identical except at \sim -0.5 eV and \sim 4.5 eV where the integrated DOS of the deep and hyperdeep defect states of $In_2Pb_{30}Te_{32}$ are larger than those of $InPb_{31}Te_{32}$ by a factor of 2. The Fermi energy is at the top of the upper localized band formed by the deep defect state.

TABLE I. Characteristics of the hyperdeep and deep defect states in PbTe with In impurity. Results obtained in PAW calculations using two different supercell sizes, $(2 \times 2 \times 2)$ supercell (64 atoms/cell) and $(3 \times 3 \times 3)$ supercell (216 atoms/cell). E_{0v} and E_{0c} are respectively at the bottom of the valence band (used as the reference for the hyper-deep states) and at the bottom of the conduction band (used as the reference for the deep states). E_1 and E_2 are respectively at the bottom of the lower (upper) localized band formed by the hyperdeep (deep) defect state. W and E_{av} are respectively the width and the average position of the lower and upper localized bands. All these quantities are measured in electron volt (eV).

	$(2 \times 2 \times 2)$ supercell		$(3 \times 3 \times 3)$ supercell	
	Hyperdeep state	Deep state	Hyperdeep state	Deep state
E_{0v}, E_{0c}	-0.077	4.903	-0.076	4.939
E_1	-0.340	4.065	-0.549	4.354
E_2	-0.663	4.719	-0.500	4.474
$W = E_2 - E_1$	0.323	0.654	0.049	0.120
$E_{\rm av} = (E_2 + E_1)/2$	-0.502	4.392	-0.525	4.414

produce *p*-type PbTe whereas Te vacancies produce *n*-type PbTe, which are consistent with the earlier model calculations.^{14,15} However, in contrast to these earlier semiempirical or model studies, *ab initio* calculations predict drastic changes in the DOS near the band-gap region, which should affect the transport properties quantitatively.

As regards monovalent impurities, Poudeu *et al.*⁸ found that Na_{0.95}Pb_mSbTe_{m+2} (m=20) was a *p*-type semiconductor with high thermoelectric FOM, ZT_{max} of ~1.7 at 650 K. The observed *p*-type behavior is consistent with our *ab initio* calculations and what is known about Na and Li impurities in PbTe. To approach a Na concentration closer to that for m=20 compound, we add one more Na atom at the middle of the 64-atom super cell. The calculation shows that the DOS in the region of the band gap does not change by this addition. Thus one can treat the Na_{0.95}Pb₂₀SbTe₂₂ as basically PbTe doped with holes. Its thermopower should be comparable to that of *p*-doped PbTe.

The behavior of trivalent impurities such as In, Ga, and Tl has been both interesting and puzzling. Gelbstein *et al.*³⁴ have recently argued that the thermoelectric property of Indoped PbTe can be understood in terms of the existence of deep-lying states generated by In-doping. We find that In not only introduces deep defect states in the band gap but also strongly reduces the DOS near the top of the valence band. This results in the reduction in the minority (hole) contribution to thermopower both due to DOS reduction and annihilation of the holes by electrons occupying the localized state in the gap. The net effect can be an increased *n*-type thermopower.

As, Sb, and Bi give rise to resonant states near the bottom of the conduction band. Although this resonant state may not take part directly in transport properties, it indirectly affects the transport properties by increasing the DOS near the bottom of the conduction band. This was pointed out in Figs. 9 and 10. The increased DOS can in principle give rise to an increased thermopower in *n*-type systems. A simple calculation³⁵ of thermopower by assuming energyindependent relaxation time showed that room temperature thermopower of AgSbPb_{2n-2}Te_{2n} can be large, as seen experimetally.⁷ However, detailed calculations of transport coefficients taking into account the changes in band structure (caused by the Sb-induced resonant states) and energy-dependent scattering indicates that this enhancement of thermopower is not true in general but depends on the carrier concentration and temperature range studied.³⁶

V. SUMMARY

In summary, our *ab initio* electronic structure calculations in $RPb_{2n-1}Te_{2n}$ (*n*=16), where *R* is a vacancy, or monovalent, divalent, or trivalent atom, show that when a Pb atom is substituted by *R*, the DOS gets perturbed over the entire valence and conduction bands. There are major changes in the DOS near the band gap region for most *R*. This should have significant impact on the transport properties of these compounds. We find that Na does not change the DOS within 0.5 eV of the band maxima; thus it is an ideal acceptor. In contrast, other alkali atoms and Ag and Cu give rise to an increase in the DOS near the top of the valence band and negligible change in the DOS near the bottom of the conduc-



FIG. 16. The total DOS of PbTe with In impurity obtained in PAW calculations using two different supercell sizes, $(2 \times 2 \times 2)$ supercell (InPb₃₁Te₃₂) and $(3 \times 3 \times 3)$ supercell (InPb₁₀₇Te₁₀₈); the distance between the two In impurities being 13.1 Å and 19.65 Å, respectively. The Fermi energy is at the top of the upper localized band formed by the deep defect state.

tion band. Hg, Cd, and Zn give rise to strong resonant state near the bottom of the conduction band and suppress the DOS near the top of the valence band (which is good for *n*-type thermoelectrics). Group V (As, Sb, and Bi) impurities introduce resonant states near the bottom of the conduction band and should be good for *n*-type thermoelectrics. In contrast, the trivalent impurities (Ga, In, and Tl) either introduce bound states in the gap or resonant states near the top of the valence band. Te vacancy also has a strong effect on the DOS of PbTe near the band gap region. Additional states appear in the band gap below the CBM; these states comprise primarily the *p* states of Pb neighboring the vacancy. In contrast, the Pb vacancy increases the DOS near the top of the valence band but by a small amount. Te vacancy should be *n* type whereas Pb vacancy should be *p* type. Iodine appears to be an ideal donor; it does not change the DOS of PbTe near the band gap, just shifts the Fermi energy to the conduction band. The divalent atoms S and Se also do not change the DOS near the CBM. There is, however, some depletion of the DOS near the VBM. These results should have important implications in the thermoelectric properties of, e.g., the *n*-type ternary compound $PbTe_{1-x}S_x$,³⁷ where one can synthesize mixed Te/S systems without appreciably affecting the conduction band states and hence the *n*-type charge and energy transport.

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